

for 5 h at room temperature. The resulting suspension was filtered, poured into H₂O, and extracted with ether. The organic phase was washed with H₂O and brine and dried. Removal of the solvent in vacuo afforded 383 mg (85%) of lactone 3. After VPC purification on column A, 3 had the following spectral data: IR 2940 (s), 2870 (s), 1780 (s), 1455 (m), 1375 (m), 1160 (m), 1090 (m), 1030 cm⁻¹ (m); NMR (60 MHz) δ 0.80–2.58 (m, 14 H), 3.95–4.40 (m, 2 H); for 220-MHz NMR data see Table I.

4-Cyclohexyldihydro-2(3H)-furanone (2). A suspension containing 15 mg (0.90 mmol) of lactone 8, 6 mg of palladium on carbon (10%), and 6 ml of MeOH was stirred under a H₂ atmosphere for 4 h at room temperature. The suspension was then filtered, poured into H₂O, and extracted with ether. The organic phase was washed with H₂O and brine and dried. Partial removal of solvent in vacuo afforded an oily residue from which lactone 2 was isolated by preparative VPC employing column B. Lactone 2 had the following spectral data: IR 2940 (s), 2860 (s), 1780 (s), 1455 (m), 1175 (s), 1050 (m), 1020 cm⁻¹ (s); for 220-MHz NMR data see Table I.

Photolysis of 2(5H)-Furanone (1) in Cyclohexane. A solution of 220 mg (2.62 mmol) of lactone 1¹³ in 250 ml of cyclohexane was flushed with N₂ for 20 min and then irradiated through Corex for 7 h under nitrogen. The photolysate was then concentrated in vacuo to afford 261 mg of an oily liquid which contained 2 and 3 in 13 and 16% yield, respectively. After VPC separation on column A, 2 and 3 were identical in all respects (e.g., VPC retention time, IR, 220-MHz NMR) with the authentic samples prepared above.

Photolysis of 2(5H)-Furanone (1) in Cyclohexane-*d*₁₂. A solution containing 15 mg of lactone 1 and 5g of cyclohexane-*d*₁₂ was placed in a quartz test tube (1 × 20 cm) fitted with a nitrogen inlet. The solution was flushed with nitrogen for 30 min and then irradiated through Corex for 20 h under nitrogen. After 20 h, the progress of the reaction was monitored by VPC on column C; approximately 80% of 1 was consumed. To the photolysis mixture was added an additional 15 mg of 1, and the mixture was irradiated for 20 h and then monitored. This process was continued until 480 mg of 1 had been destroyed. At this point the excess solvent was removed by distillation and the residue purified by VPC to yield (400–600 μ g) 2-*d* and 3-*d*. The deuterium incorporation as determined by Fourier transform 220-MHz NMR is given in Table I. Model studies with α -deuterio- α -methyl- γ -butyrolactone indicate that deuterium was not lost during purification. Examination of the recovered solvent by NMR revealed negligible hydrogen incorporation.

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Registry No.—1, 497-23-4; 7, 21681-63-0; 8, 30088-97-2; butyrolactone, 96-48-0; cyclohexanone, 108-94-1; cyclohexane-*d*₁₂, 1735-17-7.

References and Notes

- W. C. Agosta and A. B. Smith, III, *J. Am. Chem. Soc.*, **93**, 5513 (1971); S. Wolff, W. L. Schreiber, A. B. Smith, III, and W. C. Agosta, *ibid.*, **94**, 7797 (1972), and references cited therein.
- Interestingly Yoshida and Kimura observed only the β adduct when cyclopentenone was irradiated in ether solvents; Z. Yoshida and M. Kimura, *Tetrahedron*, **31**, 221 (1975).
- D. Bellus, D. R. Kearns, and K. Schaffner, *Helv. Chim. Acta*, **52**, 971 (1969); R. Reinfried, D. Bellus, and K. Schaffner, *ibid.*, **54**, 1517 (1971).
- In addition to α and β solvent adducts arising via either hydrogen abstraction by the carbonyl group or by the β -carbon atom, there are two additional mechanistically distinct modes of solvent addition in α,β -unsaturated ketones. The first involves addition of hydroxylic solvents to yield β -alkoxy ethers, presumably involving a polar 1,4-addition process; see, for example, B. J. Ramey and P. D. Gardner, *J. Am. Chem. Soc.*, **89**, 3949 (1967); P. de Mayo and J. S. Wasson, *Chem. Commun.*, 970 (1967); G. Bozzato, K. Schaffner, and O. Jeger, *Chimia*, **20**, 114 (1966); T. Matsuura and K. Ogura, *J. Am. Chem. Soc.*, **88**, 2602 (1966); O. L. Chapman, J. B. Sieja, and W. J. Weistead, Jr., *ibid.*, **88**, 161 (1966); W. G. Dauben, G. W. Shoffer, and N. D. Vietmeyer, *J. Org. Chem.*, **33**, 4060 (1968). The second mode, also leading to β -addition, involves the reaction of ground-state ketone with radicals derived from solvent through hydrogen abstraction initiated by a triplet sensitizer. This process, termed chemical sensitization, was discovered by Schenck: G. O. Schenck, G. Koltzenberg, and H. Grossman, *Angew. Chem.*, **69**, 177 (1957). For additional examples see ref 1 and R. Dulou, M. Vilkas, and M. Pfau, *C. R. Acad. Sci.*, **249**, 429 (1959); B. Fraser-Reid, D. R. Hicks, D. L. Walker, D. E. Iley, M. B. Yunker, S. K-Y. Tam, R. C. Anderson, and J. Saunders, *Tetrahedron Lett.*, 297 (1975); D. R. Hicks, R. C. Anderson, and B. Fraser-Reid, *Synth. Commun.*, **6**, 417 (1976); G. L. Bundy, *Tetrahedron Lett.*, 1957 (1975); and references cited therein.

- For related intramolecular hydrogen abstraction initiated by the β -carbon see in addition to ref 1 (a) W. Herz and M. G. Nair, *J. Am. Chem. Soc.*, **89**, 5474 (1967); (b) S. Wolff and W. C. Agosta, *J. Chem. Soc., Chem. Commun.*, 502 (1973); (c) A. B. Smith, III, and W. C. Agosta, *J. Org. Chem.*, **37**, 1259 (1972); (d) J. Gloor, G. Bernardinelli, R. Gerdl, and K. Schaffner, *Helv. Chim. Acta*, **56**, 2520 (1973); (e) F. Marti, H. Wehrli, and O. Jeger, *ibid.*, **56**, 2698 (1973); (f) J. P. Pête and J. L. Wolfhugel, *Tetrahedron Lett.*, 4637 (1973); and references cited therein. For intramolecular hydrogen atom abstraction initiated by the α -carbon atom see T. Kobayashi, M. Kurono, H. Sato, and K. Nakanishi, *J. Am. Chem. Soc.*, **94**, 2863 (1972); J. Gloor and K. Schaffner, *Helv. Chim. Acta*, **57**, 1815 (1974); M. Karuar, F. Marti, H. Wehrli, K. Schaffner, and O. Jeger, *ibid.*, **57**, 1851 (1974).
- A. B. Smith, III, and W. C. Agosta, *J. Am. Chem. Soc.*, **96**, 3289 (1974); **95**, 1961 (1973).
- S. Mageti and T. W. Gibson, *Tetrahedron Lett.*, 4889 (1973).
- This solvent adduct was shown to arise via a radical chain process initiated by hydrogen abstraction by the carbonyl oxygen: K. Ohga and T. Matsuo, *J. Org. Chem.*, **39**, 106 (1974).
- E. F. Ullman and N. Baumann, *J. Am. Chem. Soc.*, **92**, 5892 (1970).
- M. D. Shetlar, *J. Chem. Soc., Chem. Commun.*, 653 (1975).
- T. Hasegawa, H. Aoyama, and Y. Omote, *Tetrahedron Lett.*, 1901 (1975).
- The authors wish to express their gratitude to Professor Thomas W. Flechtner for a preprint of the preceding article and for the opportunity to publish concurrently. See T. W. Flechtner, *J. Org. Chem.*, preceding paper in this issue.
- M. Frank-Neumann and C. Berger, *Bull. Soc. Chim. Fr.*, 4067 (1968).
- K. Ohga and T. Matsuo, *Bull. Chem. Soc. Jpn.*, **43**, 3505 (1970); for the photoannulation of 1 with cycloalkenes see M. Tada, T. Kokubo, and T. Sato, *Tetrahedron*, **28**, 2121 (1972).
- L. M. Jackman and S. Sternhell, "Application of Nuclear Magnetic Resonance Spectroscopy in Organic Chemistry", 2d ed, Pergamon Press, Oxford, 1969, pp 23–236, and references cited therein.
- For precursor preparations of 2 and 3 see respectively E. R. Blout and R. C. Elderfield, *J. Org. Chem.*, **8**, 29 (1943); W. Reppe, *Justus Liebigs Ann. Chem.*, **596**, 158 (1955).
- V. M. Dashunin, R. V. Maeva, G. A. Kazaletova, and V. N. Belov, *Zh. Obshch. Khim.*, **34**, 3096 (1964); *Chem. Abstr.*, **61**, 15984h (1964).
- R. G. Linville and R. C. Elderfield, *J. Org. Chem.*, **6**, 270 (1941).
- Control experiments, employing these conditions, indicated that 2 and 3 were formed in the same yield and ratio as previously observed.
- R. G. Brownlee and R. M. Silverstein, *Anal. Chem.*, **40**, 2077 (1968).
- R. B. Woodward and R. Hoffman, "The Conservation of Orbital Symmetry", Academic Press, New York, N.Y., 1970.
- R. Simonaitis and J. N. Pitts, Jr., *J. Am. Chem. Soc.*, **91**, 108 (1969); **90**, 1389 (1968); R. S. Givens and W. F. Oettle, *J. Org. Chem.*, **37**, 4325 (1972).
- R. A. Cormier and W. C. Agosta, *J. Am. Chem. Soc.*, **96**, 618 (1974).

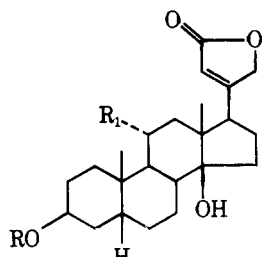
Steroids and Related Natural Products. 94. Synthesis of Toad Venom Cardenolides¹

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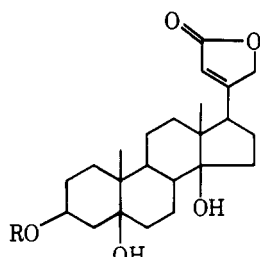
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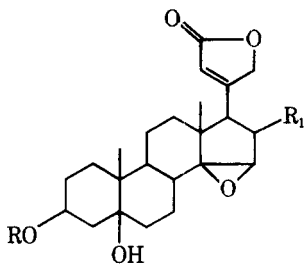
Some species of the milkweed butterfly family (Danaiidae) have been found by Reichstein and colleagues to contain cardenolides.³ The occurrence of such cardiac active plant constituents in these particular butterflies has been nicely correlated with their feeding habits which involve certain cardenolide containing plants (e.g., from the Asclepiadaceae family) and their need for an exogenous source of defensive substances. In 1970, Meyer and colleagues⁴ reported the presence of seven cardenolides in the Chinese toad venom preparation Ch'an Su. The constituents included digitoxigenin (1a), sarmentogenin (1b), periplogenin (2a), and two previously unknown 14,15 β -epoxycardenolides (3a and 3b). Whether such cardenolides represent a normal biosynthetic pathway in venom production characteristic of certain amphibians of the Bufonidae family or instead are initially obtained by ingestion of Asclepiadaceae-type plant eating insects poses an interesting biochemical question. However, the discovery^{4a} of two cardenolides bearing suberic acid ester groups (e.g., 1c) in Ch'an Su and the more recent isolation⁵ of sarmentogenin (1b), 3-suberoylarginine, and 3-pimeloylarginine esters from the skin of *Bufo vulgaris formosus*



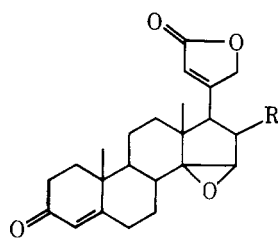
1a, R = R₁ = H
 b, R = H; R₁ = OH
 c, R = CO(CH₂)₆CO₂H; R₁ = H



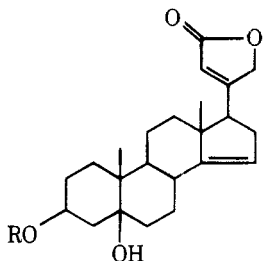
2a, R = H
 b, R = COCH₃



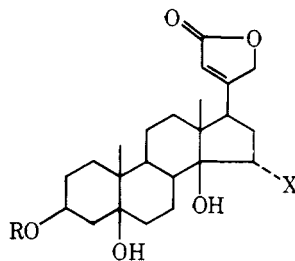
3a, R = R₁ = H
 b, R = H; R₁ = OCOCH₃
 c, R = COCH₃; R₁ = H



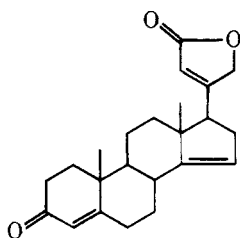
4a, R = H
 b, R = OCOCH₃



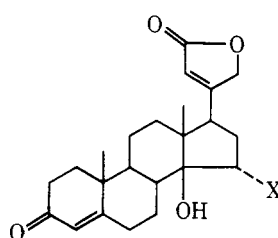
5a, R = H
 b, R = COCH₃



6a, R = H; X = I
 b, R = H; X = Br
 c, R = COCH₃; X = I
 d, R = COCH₃; X = Br



7



8a, X = I
 b, X = Br

Boulenger suggests that cardenolide formation may reflect a normal biosynthetic avenue in toads of the genus *Bufo*.

After isolation of epoxycardenolides **3a** and **3b**, the Meyer group nicely assigned structures based on spectral evidence and analogous study of the compounds (**4a** and **4b**) resulting from a dehydration-oxidation sequence. In order to extend our cytotoxicity and antineoplastic evaluations of amphibian venom constituents we have completed a formal total synthesis of the toad venom cardenolide **3a** using periplogenin (**2a**) as relay. Our earlier synthesis of periplogenin (**2a**) from digitoxigenin (**1a**) was repeated to obtain sufficient starting material.⁶ The subsequent route employed for obtaining epoxycardenolide **3a** was based on a series of reactions we developed previously for syntheses of bufotalin,^{7a} marinobufotoxin,^{7b} and bufalin.^{7c} Thus, periplogenin (**2a**) was selectively dehydrated^{7b} to 14-olefin **5a** which was converted^{7c} to halohydrins **6a** and **6b**. Periplogenin acetate (**2b**) was analogously transformed via olefin **5b** to halohydrins **6c** and

6d. Treatment of the halohydrins **6a** and **6b** or **6c** and **6d** with pyridine or basic aluminum oxide readily provided, respectively, the Ch'an Su constituent **3a** and acetate derivative **3c**. Epoxycardenolide **3a** was found to be identical with an authentic specimen obtained by the Meyer group⁴ from Ch'an Su.

The stereochemical course of halohydrin addition to olefin **5** was conclusively established by selective reduction (Raney nickel)^{7a} of the halohydrins represented by structure **6** to yield exclusively periplogenin (**2a**) and the corresponding acetate derivative **2b**. The very dependable stereochemical course of halohydrin reaction in this series was further demonstrated by conversion of 14-dehydrocanarigenone (**7**)⁶ to the 3-oxo-4-ene **4a** by way of halohydrin **8**. Ketone **4a** was found identical with a specimen obtained by selective oxidation of epoxycardenolide **3a** followed by dehydration catalyzed by Amberlite CG-120 (H⁺).

Experimental Section

The general experimental techniques in this study have been summarized in parts 93¹ and 91⁶ of this series. The same procedures have been utilized for column and thin layer chromatography (on silica gel) and establishing the mutual identity of comparison specimens (e.g., infrared spectra in KBr).

14-Dehydroperiplogenin (5a). A mixture prepared from periplogenin (**2a**, 0.25 g), methanol (45 ml), and 35% hydrochloric acid (0.05 ml) was heated at reflux for 1.5 h, poured into ice-water, and extracted with chloroform. The solvent extract was washed with water and evaporated to dryness. The crude product was column chromatographed and the fraction eluted by *n*-hexane-acetone (5:1) was recrystallized from acetone-*n*-hexane to give 14-dehydroperiplogenin (**5a**, 0.13 g) as needles: mp 200–202 °C; λ_{max} (MeOH) nm (log ε) 217 (4.20); ν_{max} (KBr) 3500 (OH), 1798, 1777, 1726 (butenolide ring), 1630, 1623 (C=C), 1445, 1030, 898, 695 cm⁻¹; ¹H NMR (10% solution in CDCl₃) δ 0.86 (3 H, s, 18-CH₃), 0.98 (3 H, s, 19-CH₃), 4.17 (1 H, broad s, 3α H), 4.76 (2 H, t, *J* = 2 Hz, 21-CH₂), 5.23 (broad s, 15-H), 5.89 (1 H, t, *J* = 2 Hz, 22-H); mass spectrum *m/e* 372 (M⁺), 354 (M⁺ - H₂O), 336 (M⁺ - 2H₂O).

Anal. Calcd for C₂₃H₃₂O₄: C, 74.16; H, 8.66. Found: C, 74.36; H, 8.70.

A 20-mg specimen of 14-dehydroperiplogenin was acetylated at room temperature with acetic anhydride (0.4 ml)-pyridine (0.28 ml) to give 14-dehydroperiplogenin acetate (**5b**, 17 mg) as needles melting at 195–198 °C (from acetone-*n*-hexane); λ_{max} (MeOH) nm (log ε) 217 (4.19); ν_{max} (KBr) 3480 (OH), 1784, 1752, 1730 (butenolide ring and ester CO), 1700, 1644, 1631 (C=C), 1450, 1255, 1245, 1240 (ester C-O), 1030, 890, 692 cm⁻¹; ¹H NMR (10% solution in CDCl₃) δ 0.86 (3 H, s, 18-CH₃), 1.01 (3 H, s, 19-CH₃), 2.09 (3 H, s, 3-OCOCH₃), 4.75 (2 H, t, *J* = 2 Hz, 21-CH₂), 5.05 (1 H, broad s, 3α H), 5.23 (1 H, broad s, 15α H), 5.88 (1 H, t, *J* = 2 Hz, 22-H); mass spectrum *m/e* 414 (M⁺), 396 (M⁺ - H₂O), 354 (M⁺ - AcOH), 336 (M⁺ - AcOH - H₂O).

Anal. Calcd for C₂₅H₃₄O₅: C, 72.43; H, 8.27. Found: C, 72.39; H, 8.25.

Synthesis of 3β,5β-Dihydroxy-14,15β-epoxycard-20(22)-enolide (3a). Method A. A solution of *N*-iodosuccinimide (30 mg) in acetone (3 ml)-water (3 ml) was added to 14-dehydroperiplogenin (**5a**, 30 mg) in acetone (4.5 ml). The mixture was stirred for 22 h at room temperature and a solution prepared from sodium sulfite (30 mg) and water (0.6 ml) was added. The solution was concentrated (to about one-third volume), poured into ice-water with stirring, and extracted with chloroform. The combined extract was washed with water, solvent was evaporated, and the crude iodohydrin (**6a**, 26 mg) was stirred in pyridine (2 ml) for 4 h at room temperature. Following removal of solvent the product was column chromatographed and the fraction eluted with *n*-hexane-acetone (5:1) was recrystallized from ethyl acetate-*n*-hexane to give 3β,5β-dihydroxy-14,15β-epoxycard-20(22)-enolide (**3a**, 21 mg) as prisms melting at 217–220 °C.

When a 15-mg sample of the crude iodohydrin (**6a**, 15 mg), obtained as described above, was chromatographed on basic alumina with benzene-chloroform (19:1-9:1), the 14β,15β-epoxide (**3a**, 8.2 mg) was, after recrystallization, isolated as prisms melting at 217–219 °C.

Method B. Substitution of *N*-bromosuccinimide (15 mg) for *N*-iodosuccinimide in the method A reaction sequence with olefin **5a** (15 mg) led to 14 mg of the crude bromohydrin (**6b**). Conversion of the bromohydrin to 14β,15β-epoxide **3a** with pyridine provided an 8.4-mg (mp 216–219 °C) yield.

Method C. When *N*-bromoacetamide (15 mg) was substituted for *N*-iodosuccinimide or *N*-bromosuccinimide as described in method A or method B, olefin 5a (15 mg) led to 16 mg of the crude bromohydrin (6b). Similar conversion of bromohydrin 6b to 14 β ,15 β -epoxide 3a by use of pyridine as described in method A provided 8.7 mg of 14 β ,15 β -epoxide 3a (mp 217–220 °C): λ_{\max} (MeOH) nm (log ϵ) 214 (4.19); ν_{\max} (KBr) 3500 (OH), 3110, 3050 (CH), 1787, 1746 (butenolide ring), 1625 (C=C), 1445, 1170, 1135, 1030, 899, 697 cm^{-1} ; $^1\text{H NMR}$ (10% solution in CDCl_3) δ 0.95 (3 H, s, 18- CH_3), 1.00 (3 H, s, 19- CH_3) 3.47 (1 H, broad s, 15 α -H), 4.16 (1 H, broad s, 3 α -H), 4.76 (2 H, t, J = 2 Hz, 21- CH_2), 5.89 (1 H, t, J = 2 Hz, 22-H); mass spectrum m/e 388 (M^+), 370 ($\text{M}^+ - \text{H}_2\text{O}$), 352 ($\text{M}^+ - 2\text{H}_2\text{O}$).

Anal. Calcd for $\text{C}_{23}\text{H}_{32}\text{O}_5$: C, 71.10; H, 8.30. Found: C, 71.23; H, 8.22.

The samples of 3 β ,5-dihydroxy-14,15 β -epoxy-5 β ,14 β -card-20(22)-enolide (3a) prepared by methods A–C were found to be identical with an authentic sample of the natural product (mp 200–221 °C, provided by Professor K. Meyer).

Synthesis of 3 β -Acetoxy-5 β -hydroxy-14,15 β -epoxycard-20(22)-enolide (3c). **Method A.** Reaction of 14-dehydroperiplogenin acetate (5b, 25 mg) with hypiodous acid prepared from *N*-iodosuccinimide (25 mg) was performed as described above for the preparation of iodohydrin 6a. After treatment with pyridine and chromatographic purification (elution with 7:1 *n*-hexane–acetone and recrystallization from ethyl acetate–*n*-hexane) the crude iodohydrin acetate (6c, 23 mg) gave rise to 3 β -acetoxy-5 β -hydroxy-14,15 β -epoxycard-20(22)-enolide (3c, 18 mg) as prisms melting at 217–220 °C.

Method B. The preceding reaction was repeated using 15 mg of olefin acetate 5b and 15 mg of *N*-bromoacetamide. Alumina (basic) chromatographic treatment of the crude bromohydrin (6d, 14 mg) with benzene–chloroform (19:1) as eluent provided 7.8 mg of 14 β ,15 β -epoxy acetate 3c melting at 216–220 °C.

Method C. Alcohol 3a (10 mg) was acetylated with acetic anhydride (0.014 ml)–pyridine (0.02 ml) and the product was isolated by column chromatography as described in method A. By this means 7 mg of 14 β ,15 β -epoxy acetate 3c was obtained which melted at 218–220 °C and was identical with the sample prepared by method A or method B; λ_{\max} (MeOH) nm (log ϵ) 213 (4.18); ν_{\max} (KBr) 3480 (OH), 3100, 3048 (CH), 1785, 1750, 1728 (butenolide ring and ester CO), 1700, 1642, 1626 (C=C), 1445, 1250, 1240 (ester C–O), 1170, 1135, 1030, 897, 695 cm^{-1} ; $^1\text{H NMR}$ (10% solution in CDCl_3) δ 0.95 (3 H, s, 18- CH_3), 1.02 (3 H, s, 19- CH_3), 2.08 (3 H, s, 3- OCOCH_3), 3.47 (1 H, broad s, 15 α -H), 4.77 (2 H, a narrow quartet, J = 2.5 and 1.5 Hz, 21- CH_2), 5.24 (1 H, broad s, 3 α -H), 5.88 (1 H, t, J = 2.5 Hz, 22-H); mass spectrum m/e 430 (M^+), 412 ($\text{M}^+ - \text{H}_2\text{O}$), 394 ($\text{M}^+ - 2\text{H}_2\text{O}$), 370 ($\text{M}^+ - \text{AcOH}$).

Anal. Calcd for $\text{C}_{25}\text{H}_{34}\text{O}_6$: C, 69.74; H, 7.96. Found: C, 69.77; H, 7.89.

Periplogenin (2a). The crude iodohydrin (6a, 21 mg) prepared from 14-dehydroperiplogenin (5a, 22 mg) and *N*-iodosuccinimide (22 mg) was treated^{7c} with freshly prepared Raney nickel (approximately 0.8 g) for 4 h at 18–20 °C in a nitrogen atmosphere. The solution was filtered and the filtrate was concentrated to provide an 18-mg residue, which was subjected to column chromatography. The fraction eluted by 1:3 *n*-hexane–acetone was recrystallized from methanol to afford 11 mg of periplogenin (2a) melting at 226–232 °C.

By an analogous reduction reaction the crude bromohydrin (6b, 10 mg) prepared from 14-dehydroperiplogenin (5a, 12 mg) and *N*-bromoacetamide (12 mg) was treated (under nitrogen) with freshly prepared Raney nickel (ca. 0.5 g). By the same isolation procedure, 5 mg of periplogenin (2a, mp 226–231 °C) was obtained. The specimens of periplogenin (2a) obtained by both procedures were found identical with natural periplogenin.

Periplogenin Acetate (2b). A sample of the crude iodohydrin acetate (6c, 18 mg) prepared from 14-dehydroperiplogenin acetate (5b, 20 mg) and *N*-iodosuccinimide (20 mg) was converted to periplogenin acetate (2b, 10 mg, mp 230–238 °C) using Raney nickel (about 0.5 g, 4 h at 18 °C) as summarized above for obtaining periplogenin (2a).

With the same nickel (approximately 0.3 g) procedure, 8 mg of the crude bromohydrin acetate (6d, obtained using *N*-bromosuccinimide) provided 3.8 mg of periplogenin acetate (2b) melting at 230–237 °C.

Both samples of periplogenin acetate (2b) were found identical with an authentic sample.

3-Oxo-14,15 β -epoxycarda-4,20(22)-dienolide (4a). **Method A.** A solution of *N*-iodosuccinimide (30 mg) in acetone (3 ml)–water (3 ml) was added to 30 mg of 14-dehydrocarnigenone [3-oxocarda-4,14,20(22)-trienolide 7] in acetone (4.5 ml). The remaining reaction sequence and isolation procedure was completed (except for 18 h with

sodium sulfite and 3 h with pyridine) as described above for obtaining epoxycardenolide 3a. Recrystallization from ethyl acetate–*n*-hexane afforded 20 mg of 3-oxo-14,15 β -epoxy-14 β -carda-4,20(22)-dienolide (4a) as prisms melting at 236–240 °C. The product was identical with the sample prepared below from 3 β ,5 β -dihydroxy-14,15 β -epoxycard-20(22)-enolide (3a).

Method B. The preceding reaction was repeated using 15 mg of olefin 7 and 15 mg of *N*-bromoacetamide. Analogous treatment of the crude bromohydrin (8b, 14 mg) with pyridine provided epoxide 4a (6.6 mg) melting at 236–239 °C which was identical with the sample prepared below from diol 3a.

When the crude bromohydrin (8b, 10 mg) obtained by similar treatment of olefin 7 (12 mg) with *N*-bromosuccinimide (12 mg), was chromatographed in benzene–chloroform (14:1) on basic alumina 4.7 mg of epoxide 4a, mp 237–240°C, was isolated.

Method C (From 3 β ,5 β -Dihydroxy-14,15 β -epoxycard-20(22)-enolide, 3a). Epoxy alcohol 3a (18 mg) in pyridine (0.48 ml) was oxidized (room temperature, 16 h) with chromium trioxide (17 mg)–pyridine (0.18 ml) complex. Excess reagent was removed with methanol and the mixture was poured into ice–water and extracted with chloroform. The combined extract was washed with water and concentrated to dryness. The crude epoxy ketone (16 mg) was employed in the following dehydration reaction without further purification.

A mixture prepared from the epoxy ketone (15 mg), 0.15 g of Amberlite CG-120 (H^+ form), and methanol (1.5 ml) was stirred at room temperature for 8 h. The solution was filtered and concentrated to dryness and the crude product was purified by column chromatography. The fraction eluted with 9:1 *n*-hexane–acetone was recrystallized from ethyl acetate–*n*-hexane to yield 9.2 mg of epoxide 4a melting at 237–241 °C (lit^{7b} mp 237–248 °C); λ_{\max} (MeOH) nm (log ϵ) 227–230 (4.33); ν_{\max} (KBr) 3100, 3048, (CH), 1780, 1735, 1715 (butenolide ring and saturated ketone), 1700, 1623 (C=C), 1445, 1170, 1120, 1022, 890, 858, 780, 750 cm^{-1} ; $^1\text{H NMR}$ (10% solution in CDCl_3) δ 1.01 (3 H, s, 18- CH_3), 1.26 (3 H, s, 19- CH_3), 3.45 (1 H, broad s, 15 α H), 4.78 (2 H, t, J = 2 Hz, 21- CH_2), 5.73 (1 H, broad s, 4-H), 5.88 (1 H, t, J = 2 Hz, 22-H); mass spectrum m/e 368 (M^+), 350 ($\text{M}^+ - \text{H}_2\text{O}$).

Anal. Calcd for $\text{C}_{23}\text{H}_{28}\text{O}_4$: C, 74.95; H, 7.66. Found: C, 74.99; H, 7.63.

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Registry No.—2a, 514-39-6; 2b, 13077-88-8; 3a, 31655-31-9; 3c, 31655-36-4; 4a, 24366-48-1; 5a, 60967-71-7; 5b, 60967-72-8; 6a, 60967-73-9; 6b, 60967-74-0; 6c, 60967-75-1; 6d, 60967-76-2; 7, 19637-08-2; 8, 24366-46-9; *N*-iodosuccinimide, 516-12-1; *N*-bromosuccinimide, 128-08-5; *N*-bromoacetamide, 79-15-2.

References and Notes

- (1) For contribution 93 refer to Y. Kamano, G. R. Pettit, M. Tozawa, Y. Komeichi, and M. Inoue, *J. Org. Chem.*, **40**, 2136 (1975).
- (2) (a) Department of Chemistry, School of Medicine, Premedical Course, The Jikei University, Tokyo, Japan; (b) Department of Biochemical Pharmacology, Faculty of Pharmaceutical Sciences, University of Chiba, Chiba, Japan.
- (3) J. von Euv, L. Fiskelson, J. A. Parsons, T. Reichstein, and M. Rothschild, *Nature (London)*, **214**, 35 (1967); L. P. Brower, P. B. McEvoy, K. L. Williamson, and M. A. Flannery, *Science*, **177**, 426 (1972).
- (4) (a) N. Höriger, H. H. A. Linde, and K. Meyer, *Helv. Chim. Acta*, **53**, 1503 (1970); (b) N. Höriger, D. Zivanov, H. H. A. Linde, and K. Meyer, *ibid.*, **53**, 2051 (1970).
- (5) Y. Fujii, K. Shimada, Y. Niizaki, and T. Nambara, *Tetrahedron Lett.*, 3017 (1975).
- (6) Y. Kamano, G. R. Pettit, and M. Tozawa, *J. Chem. Soc., Perkin Trans. 1*, 1976 (1975); Y. Kamano, G. R. Pettit, and M. Tozawa, *J. Org. Chem.*, **39**, 2319 (1974).
- (7) (a) Y. Kamano, G. R. Pettit, and M. Inoue, *J. Org. Chem.*, **39**, 3007 (1974); (b) G. R. Pettit and Y. Kamano, *ibid.*, **39**, 3003 (1974); (c) Y. Kamano and G. R. Pettit, *ibid.*, **38**, 2202 (1973). See also M. Heller, F. J. McEvoy, and S. Bernstein, *Steroids*, **3**, 193 (1964); C. R. Engel and G. Bach, *ibid.*, **3**, 593 (1964).